

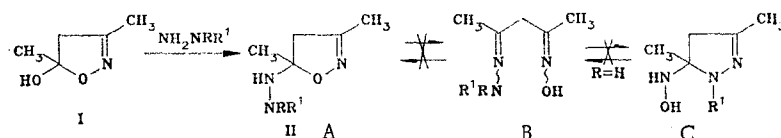
5. J. Elguero and R. Jacquier, Bull. Soc. Chim. Fr., No. 9, 2832 (1966).  
 6. R. N. Butler and I. P. James, J. Chem. Soc., Perkin Trans. 1, 553 (1982).

### 5-HYDRAZINO-2-ISOXAZOLINES

K. N. Zelenin, A. Yu. Ershov,  
 I. P. Bezhan, V. A. Khrustalev,  
 and S. I. Yakimovich

UDC 547.786.1.07

In view of the high and varied biological activity of many isoxazole derivatives [1], to find new methods for their synthesis is becoming timey. We have established that a simple method of obtaining some 5-hydrazino-2-isoxazolines IIa-c is the reaction of the readily available 5-hydroxy-3,5-dimethyl-2-isoxazolines I [2] with the respective hydrazines.



II a R=H, R<sup>1</sup>=COC<sub>6</sub>H<sub>5</sub>; b R=H, R<sup>1</sup>=COC<sub>6</sub>H<sub>4</sub>NO<sub>2</sub>-p; c R=R<sup>1</sup>=CH<sub>3</sub>

According to <sup>1</sup>H and <sup>13</sup>C NMR data, the compounds IIa-c have the cyclic structure A, and show no tendency to go over in various solvents to the linear form B. Incidentally the hydrazinooximes of 1,3-dioxo compounds are unknown.

With IIa,b the theoretical possibility of formation of a pyrazoline ring C must also be considered. But this variant must be excluded because in the <sup>13</sup>C NMR spectra, the C<sub>(5)</sub> atom of tautomer C ought to be located in the 80-90 ppm range [3]. In the case of IIc, however, which is incapable of such a transition, the C<sub>(5)</sub> signal appears at a weaker field (98.9 ppm). The value for the C<sub>(5)</sub> signal in IIa, viz., 100.2 ppm, is also evidence that it is located in a N,O-, and not a N,N environment. An attempt to synthesize a type C compound by another route, by the reaction of hydroxylamine and 1-benzoyl-5-hydroxy-3,5-dimethyl-2-pyrazoline [4] yielded benzhydrazide and a mixture of I with acetylacetone dioxime.

II was synthesized by boiling equimolar amounts of the reagents in benzene solution in the presence of KU-2 carion exchanger in the H<sup>+</sup> form. The elemental compositions of IIa-c agree with those calculated.

IIa: yield 72%, mp 107-108° (from hexane-ethyl acetate mixture). PMR spectrum (DMFA-D<sub>7</sub>): 1.69 (s, 3H), and 2.01 (t, 3H, J = 1.0 Hz) - 2 CH<sub>3</sub> groups; 2.99 and 3.21 (2H, J<sub>AB</sub> = 18.0, J = 1.0 Hz, 4-H); 5.85 (d, 1H, J = 6.0 Hz, NH); 7.53-8.0 (m, 5H, H<sub>arom</sub>); 9.85 ppm (d, 1H, J = 6.0 Hz, NHCO). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>): 13.6 (J<sub>CH</sub> = 129.0 Hz) and 23.6 (J<sub>CH</sub> = 129.0, J<sub>CCCH</sub> = 4.0 Hz) - 2 CH<sub>3</sub> groups; 47.0 (J<sub>CH</sub> = 136.0, J<sub>CCCH</sub> = 4.0 Hz, C<sub>(4)</sub>); 100.2 (C<sub>(5)</sub>); 158.3 (J<sub>CCCH</sub> = 4.5 Hz, C<sub>(3)</sub>); 170.2 (J<sub>CCCH</sub> = 6.8 Hz, C=O); 128.7-134.6 ppm (C<sub>arom</sub>).

IIb: yield 62%, mp 153-155° (from toluene). PMR spectrum (DMFA-D<sub>7</sub>): 1.50 (s, 3H) and 1.83 (t, 3H, J = 1.0 Hz) - 2 CH<sub>3</sub> groups; 2.92 and 3.10 (2H, J<sub>AB</sub> = 18.0, J = 1.0 Hz, 4-H); 5.93 (d, 1H, J = 5.0 Hz, NH); 8.07 (d, 2H, J = 9.0 Hz) and 8.31 (d, 2H, J = 9.0 Hz, H<sub>arom</sub>) 10.13 ppm (d, 1H, J = 5.0 Hz, NHCO).

IIc: yield 85%, mp 48-49° (from hexane-ethyl acetate mixture). PMR spectrum (DMFA-D<sub>7</sub>): 1.44 (s, 3H), and 1.92 (t, 3H, J = 1.0 Hz) - 2 CH<sub>3</sub> groups; 2.46 (s, (CH<sub>3</sub>)<sub>2</sub>N); 2.72 and 3.00 (2H, J<sub>AB</sub> = 16.0, J = 1.0 Hz, 4-H); 3.94 ppm (br. s, 1H, NH). <sup>13</sup>C NMR spectrum (CD<sub>3</sub>OD): 13.1 (J<sub>CH</sub> = 129.0 Hz) and 24.1 (J<sub>CH</sub> = 129.0, J<sub>CCCH</sub> = 4.5 Hz) - 2 CH<sub>3</sub> groups; 45.3 (J<sub>CH</sub> = 135.0, J<sub>CCCH</sub> = 3.0 Hz, C<sub>(4)</sub>); 49.5 (J<sub>CH</sub> = 132.0, J<sub>CCCH</sub> = 4.5 Hz, (CH<sub>3</sub>)<sub>2</sub>N); 98.9 (J<sub>CCCH</sub> = 3.0 Hz, C<sub>(5)</sub>), 154.9 ppm (J<sub>CCCH</sub> = 6.5 Hz, C<sub>(3)</sub>).

### LITERATURE CITED

1. S. D. Sokolov, Usp. Khim., **68**, 533 (1979).

S. M. Kirov Academy of Military Medicine, Leningrad 194175. Translated from Khimiya Geterotsiklicheskih Soedinenii, No. 6, pp. 855-856, June, 1985. Original article received January 21, 1985.

2. R. Escalé, F. Petrus, and J. Verducci, *Bull. Soc. Chim. Fr.*, Nos. 3-4, 725 (1974).
3. K. N. Zelenin, M. Yu. Malov, I. P. Bezhan, V. A. Khrustalev, and S. I. Yakimovich *Khim. Geterotsikl. Soedin.*, No. 6, 855 (1985).
4. V. G. Yusipov, S. I. Yakimovich, S. D. Nasiridinov, and N. A. Parpiev, *Zh. Org. Khim.*, 16, 415 (1980).